

Measuring and Modeling the Speed of Sound and the Viscosity of N₂O and NO.

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The speed of sound and the viscosity of nitrous oxide and nitric oxide have been accurately measured using advanced acoustic techniques [1-3]. The measurements span the temperature range 220 to 460 K, and pressures up to 3.3 MPa. In previous work, we fit speed-of-sound data with hard-core square-well and hard-core Lennard-Jones intermolecular potentials. The resulting virial equations of state are capable of predicting gas densities to better than 0.1 % at pressures up to 1.5 MPa. Here, we attempt to use these potentials and several other spherically symmetric intermolecular potentials to fit simultaneously the zero-density viscosities, $\eta_0(T)$ and the second acoustic virial coefficient $\beta_a(T)$. Our objective is to use model potentials to extrapolate $\eta_0(T)$ and $\beta_a(T)$ to much higher temperatures and to predict the thermal conductivity.

- [1] A. R. H. Goodwin and M. R. Moldover, *J. Chem. Phys.* 95, 5236 (1991).
- [2] K. A. Gillis, *Int. J. Thermophys.* 18, 73 (1997).
- [3] K. A. Gillis, J. B. Mehl, M. R. Moldover, Submitted to *J. Acous. Soc. Amer.*